the type of interaction in which one of the components is weaker than the other. Another criterion observed is that the H atom lies in or close to the plane of the donor and the two acceptor atoms. The planarity of $\mathrm{N}(1)$, $\mathrm{H}(\mathrm{N} 1), \mathrm{Cl}$ and $\mathrm{O}(2)^{a}$ for the present molecule is indicated by the sum of the three angles at $\mathrm{H}(\mathrm{N} 1)$, which is $360^{\circ}$.

The short intramolecular distance of 2.923 (9) $\AA$ between $\mathrm{N}(2)$ and $\mathrm{C}(12)$ in the molecule mentioned earlier also appears to be due to a three-center hydrogen bond involving the atoms $\mathrm{N}(2), \mathrm{O}(1)^{b}(x, y+$ $1, z$ ) and $\mathrm{C}(12)$, the weaker component being $\mathrm{N}(2)-\mathrm{H}(\mathrm{N} 2) \cdots \mathrm{C}(12)\left(83^{\circ}\right)$. The relevant distances are $\mathrm{H}(\mathrm{N} 2) \cdots \mathrm{C}(12)=2.88$ and $\mathrm{H}(\mathrm{N} 2) \cdots \mathrm{O}(1)^{b}=$ $1.96 \AA$. The angles $\mathrm{N}(2)-\mathrm{H}(\mathrm{N} 2) \cdots \mathrm{C}(12)\left(83^{\circ}\right)$, $\mathrm{N}(2)-\mathrm{H}(\mathrm{N} 2) \cdots \mathrm{O}(1) \quad\left(154^{\circ}\right)$ and $\mathrm{C}(12) \cdots \mathrm{H}(\mathrm{N} 2)$ $\cdots \mathrm{O}(1)\left(114^{\circ}\right)$ at $\mathrm{H}(\mathrm{N} 2)$ sum to $351^{\circ}$.

## References

Herbert, V. (1966). The Pharmacological Basis of Therapeutics, 3rd ed., edited by L. S. Goodman \& A. Gilman. New York: Macmillan.
International Tables for $X$-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press.
iUPAC-IUB Commission on Biochemical Nomen. clature (1970). J. Mol. Biol. 52, 1-17.
Jeffrey, G. A. \& Maluszynska, H. (1982). Int. J. Biol. Macromol. In the press.
Main, P., Hull, S. E., Lessinger, L., Germain, G.. Declerce, J. P. \& Woolfson, M. M. (1978). MULTaN 78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
Ringertz, H. (1971). Acta Cryst. B27, 285-291.
Seal, A. \& Ray, S. (1981). Indian J. Phys. 55A, 414-416.
Sequeira, A., Rajagopal, H. \& Chidambaram, R. (1972). Acta Cryst. B28, 2514-2519.

Acta Cryst. (1982). B38, 2089-2091

# $N$-Methyl-10-azabicyclo[4.3.1 ]decane-8-spiro-5'-hydantoin* $\dagger$ 

By F. Florencio, P. Smith-Verdier and S. García-Blanco<br>Departamento de Rayos X, Instituto de Química-Física 'Rocasolano', Serrano 119, Madrid-6, Spain

(Received 16 December 1981; accepted 26 February 1982)


#### Abstract

C}_{12} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{2}, M_{r}=237.303\), monoclinic, $P 2_{1} / n, \quad a=6.853(1), \quad b=11.995(1), \quad c=$ 14.835 (1) $\AA, \beta=90.58(3)^{\circ}, V=1217.40 \AA^{3}, Z=$ $4, D_{x}=1.293 \mathrm{~g} \mathrm{~cm}^{-3}, \mu(\mathrm{Mo} K \alpha)=0.8396 \mathrm{~cm}^{-1}$. The structure was solved by direct methods and refined to an $R$ of 0.042 for 2621 observed reflections. The bicyclo[4.3.1]decane system has a chair-boat conformation. The $\mathrm{CH}_{3}$ group is attached in an axial position.


Introduction. The title compound was synthesized by Dr E. Martinez Muñoz of the Departamento de Quimica Orgánica de la Facultad de Farmacia de la Universidad Complutense de Madrid. The chemical structure could not be unequivocally established by conventional spectroscopic techniques.

The crystal used for all X-ray measurements was a transparent colourless parallelepiped of approximate dimensions $0.33 \times 0.27 \times 0.21 \mathrm{~mm}$. An accurate determination of the cell parameters was made by a least-squares fit of the settings of 25 reflections on an automatic Enraf-Nonius CAD-4 diffractometer.

[^0]The space group $P 2_{1} / n$ was assigned uniquely from the systematic absences. The intensities of reflections up to $2 \theta=60^{\circ}$ were collected on the diffractometer using graphite-monochromatized Mo $K \alpha$ radiation. 3486 independent reflections were recorded with the $\theta-2 \theta$ scan mode, 2621 of which were considered as observed [ $I \geq 2 \sigma(I)$ where $\sigma(I)$ was calculated from counting statistics]. The intensities were not corrected for absorption.
The structure was solved by MULTAN80 (Main, Fiske, Hull, Lessinger, Germain, Declercq \& Woolfson, 1980). The solution was based on phases determined for 250 reflections with $E \geq 1.90$. The $E$ map with the best figure of merit revealed the positions of all the non-hydrogen atoms.
After full-matrix least-squares refinement with anisotropic temperature factors ( $R=0.090$ ), the H atoms were located in a difference map. Final refinement, in which the positional parameters and isotropic temperature factors of the H atoms were allowed to vary, converged at $R=0.042$ and $R_{w}=\left[\sum w\left(\left|F_{o}\right|-\right.\right.$ $\left.\left.\left|F_{c}\right|\right)^{2} / \sum w\left|F_{o}\right|^{2}\right]^{1 / 2}=0.047$. The function minimized was $\sum w\left(\left|F_{o}\right|-\left|F_{c}\right|\right)^{2}$ with $w=1 \cdot 00$. The final difference map contained no peaks $>0.3 \mathrm{e} \AA^{-3}$. The scattering factors were taken from International Tables for

Table 1. Atomic coordinates ( $\times 10^{5}$ for $\mathrm{C}, \mathrm{N}$ and O , $\times 10^{3}$ for H ) and thermal parameters ( $U_{\text {eq }} \times 10^{4}$ for C , N and $\mathrm{O}, U_{\text {iso }} \times 10^{3}$ for H )

|  | $U_{\text {eq }}=\frac{1}{3} \bigsqcup_{i} \beth_{j} U_{i j} a_{i}^{*} a_{j}^{*} a_{i} a_{j} \cos \left(a_{i} a_{j}\right)$. |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $\begin{gathered} U_{\mathrm{eq}} / U_{\mathrm{iso}} \\ \left(\AA^{2}\right) \end{gathered}$ |
| C(1) | 57969 (25) | 33893 (16) | 63381 (12) | 302 (5) |
| C(2) | 52199 (27) | 31784 (18) | 53433 (12) | 334 (5) |
| C(3,5') | 30700 (25) | 28133 (15) | 51980 (11) | 276 (5) |
| C(4) | 24058 (26) | 19676 (16) | 59183 (12) | 310 (5) |
| C(5) | 30996 (26) | 22384 (15) | 68857 (11) | 289 (5) |
| C(6) | 19130 (27) | 31896 (18) | 73114 (13) | 358 (6) |
| C(7) | 29650 (30) | 38625 (19) | 80445 (13) | 395 (6) |
| C(8) | 47387 (30) | 45329 (17) | 77177 (14) | 376 (6) |
| C (9) | 50471 (30) | 45121 (16) | 67026 (14) | 369 (6) |
| $\mathrm{N}(10)$ | 52328 (21) | 24478 (13) | 69215 (9) | 275 (4) |
| $\mathrm{C}(11)$ | 63472 (33) | 14266 (19) | 67113 (14) | 428 (7) |
| $\mathrm{N}\left(1^{\prime}\right)$ | 17083 (24) | 37445 (14) | 50544 (10) | 340 (5) |
| $\mathrm{C}\left(2^{\prime}\right)$ | 8715 (28) | 37261 (17) | 42239 (12) | 335 (5) |
| $\mathrm{N}\left(3^{\prime}\right)$ | 16229 (24) | 28234 (14) | 37589 (10) | 340 (5) |
| C(4') | 29156 (26) | 22264 (16) | 42765 (11) | 309 (5) |
| $\mathrm{O}(1)$ | -3333 (24) | 43780 (13) | 39146 (10) | 486 (5) |
| $\mathrm{O}(2)$ | 37621 (23) | 13878 (13) | 40607 (10) | 453 (5) |
| H(11) | 727 (3) | 342 (2) | 636 (1) | 14 (6) |
| H(21) | 607 (3) | 259 (2) | 508 (1) | 10 (5) |
| H(22) | 551 (3) | 385 (2) | 498 (1) | 11 (5) |
| H(41) | 289 (3) | 118 (2) | 572 (1) | 10 (5) |
| H(42) | 98 (3) | 191 (2) | 591 (1) | 15 (6) |
| H(51) | 282 (3) | 153 (2) | 725 (1) | 9 (5) |
| H(61) | 74 (3) | 285 (2) | 759 (1) | 18 (6) |
| H(62) | 140 (3) | 375 (2) | 684 (1) | 13 (6) |
| H(71) | 322 (3) | 337 (2) | 856 (2) | 24 (7) |
| H(72) | 203 (3) | 437 (2) | 835 (1) | 17 (6) |
| H(81) | 594 (3) | 432 (2) | 800 (1) | 14 (6) |
| H(82) | 458 (3) | 534 (2) | 789 (1) | 11 (5) |
| H(91) | 383 (3) | 477 (2) | 640 (1) | 17 (6) |
| H(92) | 594 (3) | 509 (2) | 653 (1) | 18 (6) |
| H(111) | 783 (4) | 155 (2) | 679 (2) | 45 (9) |
| $\mathrm{H}(112)$ | 621 (3) | 114 (2) | 607 (2) | 23 (6) |
| H(113) | 595 (4) | 78 (2) | 711 (2) | 26 (7) |
| $\mathrm{H}\left(1^{\prime} 1\right)$ | 137 (3) | 431 (2) | 544 (1) | 14 (6) |
| H(3'1) | 129 (4) | 267 (2) | 320 (2) | 24 (7) |

$X$-ray Crystallography (1974). The computations were made with programs of the XRAY 70 system (Stewart, Kundell \& Baldwin, 1970). Final atomic coordinates are given in Table 1.*

Discussion. The crystallographic atom numbers, bond lengths and angles are given in Fig. 1 and Table 2. Distances and angles do not deviate significantly from their expected values. The molecule contains a sixmembered (I) and a seven-membered (II) ring joined by a common $\mathrm{C}(1)-\mathrm{N}(10)-\mathrm{C}(5)$ bridge with a $\mathrm{CH}_{3}$ group

[^1]

Fig. 1. Scheme and crystallographic numbering of the molecule.

Table 2. Interatomic distances ( $\AA$ ) and angles ( ${ }^{\circ}$ )
The e.s.d.'s are in parentheses.

| $\mathrm{C}(1)-\mathrm{C}(2)$ | 1.545 (3) | $\mathrm{C}(6)-\mathrm{C}(7)$ | 1.529 (3) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(1)-\mathrm{C}(9)$ | 1.541 (3) | $\mathrm{C}(7)-\mathrm{C}(8)$ | 1.540 (3) |
| $\mathrm{C}(1)-\mathrm{N}(10)$ | 1.477 (2) | $\mathrm{C}(8)-\mathrm{C}(9)$ | 1.523 (3) |
| $\mathrm{C}(2)-\mathrm{C}\left(3.5{ }^{\prime}\right)$ | 1.550 (3) | $\mathrm{N}(10)-\mathrm{C}(11)$ | 1.478 (3) |
| $\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}(4)$ | 1.545 (3) | $\mathrm{N}\left(1^{\prime}\right) \mathrm{C}\left(2^{\prime}\right)$ | 1.354(2) |
| $\mathrm{C}\left(3.5^{\circ}\right)-\mathrm{N}\left(1^{\prime}\right)$ | 1.470 (2) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right)$ | 1.386 (3) |
| $\mathrm{C}\left(3.5^{\circ}\right)-\mathrm{C}\left(4^{\prime}\right)$ | 1.540 (3) | $\mathrm{C}\left(2^{\prime}\right)$-O(1) | 1.223 (3) |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | 1.542 (2) | $\mathrm{N}\left(3^{\prime}\right)$ - $\mathrm{C}\left(4^{\prime}\right)$ | 1.369 (2) |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | 1.540 (3) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}(2)$ | 1-206(2) |

$\mathrm{C}(9)-\mathrm{C}(1)-\mathrm{N}(10)$
$\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{N}(10)$
$\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{N}(10)$
$\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(9)$
$\begin{array}{llll}C(2)-C(1)-\mathrm{C}(9) & 113.3(2) & \mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9) & 114.9(2) \\ \mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3.5)-\mathrm{C}(8) & 111.9(1)\end{array}$
$\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}\left(3.5^{\prime}\right) \quad 114.4(2)$
$\mathrm{C}(2)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) \quad 108.0(2) \quad \mathrm{C}(5)-\mathrm{N}(10)-\mathrm{C}(11) \quad 111.9$ (1)
$\mathrm{C}(2)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right) \quad 114.0(2) \quad \mathrm{C}(1)-\mathrm{N}(10)-\mathrm{C}(11) \quad 111.8(2)$
$\begin{array}{llll}\mathrm{C}(2)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}(4) & 112.1(2) & \mathrm{C}\left(3.5^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right) & 112.5(2) \\ \mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) & 100.4(1) & \mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}(1) & 127.5(2)\end{array}$
$\begin{array}{llll}\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) & 100.4(1) & \mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}(1) & 127.5(2) \\ \mathrm{C}(4)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) & 107.2(1) & \mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right) & 108.0\end{array}$
$\begin{array}{llll}\mathrm{C}(4)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) & 107.2(1) & \mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right) & 108.0(2) \\ \mathrm{C}(4)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right) & 114.1(2) & \mathrm{N}\left(3^{\prime}\right)-\mathrm{C}\end{array}$
$\begin{array}{llll}\mathrm{C}(4)-\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right) & 114.1(2) & \mathrm{N}\left(3^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{O}(1) & 124.4 \text { (2) }\end{array}$
$\begin{array}{llll}\mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}(4)-\mathrm{C}(5) & 114.5(2) & \mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right)-\mathrm{C}(4) & 111.8(2) \\ \mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(10) & 111.3(1) & \mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{N}\left(3^{\prime}\right) & 107.3(2)\end{array}$
$\begin{array}{llll}\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6) & 112.3(2) & \mathrm{N}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}(2) & 126.7(2) \\ \mathrm{C}(6)-\mathrm{C}(5)-\mathrm{N}(10) & 112.6(2) & \mathrm{C}\left(3^{\prime} 5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}(2) & 126.0\end{array}$
$\begin{array}{llll}\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{N}(10) & 112.6(2) & \mathrm{C}\left(3.5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}(2) & 126.0(2) \\ \mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7) & 115.8(2) & \end{array}$
attached to the N atom in the axial position and a hydantoin ring substituted at the spiranic $\mathrm{C}\left(3,5^{\prime}\right)$.
Ring I has a slightly deformed chair conformation. The asymmetry parameters (Duax \& Norton, 1975), $\Delta C_{s}^{3, s^{\prime}}=0.48^{\circ}, \Delta C_{2}^{\left(2-3.5^{\prime}\right)}=8.40^{\circ}, \Delta C_{2}^{(1-2)}=17 \cdot 80^{\circ}$, show that there is an approximate $C_{s}$ plane passing through $\mathrm{C}\left(3,5^{\prime}\right)$ and $\mathrm{N}(10)$. These atoms are displaced by -0.523 (2) and 0.665 (2) $\AA$ respectively out of the plane defined by the remaining ring atoms. The seven-membered ring (II) has a boat conformation; $\mathrm{C}(1), \mathrm{C}(5), \mathrm{C}(6)$ and $\mathrm{C}(9)$ form the bottom of the boat. Torsional angles, Table 3, are in very good agreement with those of an ideal boat. The deviation of $\mathrm{N}(10)$, $\mathrm{C}(7)$ and $\mathrm{C}(8)$ from the least-squares plane through $\mathrm{C}(1), \mathrm{C}(5), \mathrm{C}(6)$ and $\mathrm{C}(9)$ are -0.718 (2), -1.153 (2) and -1.200 (2) $\AA$ respectively. The molecular geometry approximates to $m$ [mirror plane defined by $\mathrm{C}\left(3,5^{\prime}\right), \mathrm{N}(10)$ and $\left.\mathrm{C}(11)\right]$. The atoms belonging to the hydantoin ring are approximately in this plane, only $\mathrm{C}\left(2^{\prime}\right), \mathrm{O}(1)$ and $\mathrm{O}(2)$ deviate slightly. By comparing the deformation of the cyclohexane ring (I) with that found in N -methyltropane-3-spiro- $5^{\prime}$-hydantoin (Smith-

$$
\begin{equation*}
\mathrm{C}_{12} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{2} \tag{2091}
\end{equation*}
$$

Table 3. Torsional angles $\left({ }^{\circ}\right)$

| $C(1)-C(2)-C\left(3.5^{\prime}\right)-C(4)$ | $-41 \cdot 3(2)$ | $C(1)-N(10)-C(5)-C(6)$ | $-67 \cdot 6(2)$ |
| :--- | ---: | :--- | ---: |
| $C(2)-C\left(3.5^{\prime}\right)-C(4)-C(5)$ | $41.8(2)$ | $N(10)-C(5)-C(6)-C(7)$ | $-27.2(2)$ |
| $C\left(3.5^{\prime}\right)-C(4)-C(5)-N(10)$ | $-50 \cdot 9(2)$ | $C(5)-C(6)-C(7)-C(8)$ | $64.9(2)$ |
| $C(4)-C(5)-N(10)-C(1)$ | $59.5(2)$ | $C(6)-C(7)-C(8)-C(9)$ | $5.7(3)$ |
| $C(5)-N(10)-C(1)-C(2)$ | $-59.5(2)$ | $C(7)-C(8)-C(9)-C(1)$ | $-72 \cdot 3(2)$ |
| $N(10)-C(1)-C(2)-C\left(3.5^{\prime}\right)$ | $50.3(2)$ | $C(8)-C(9)-C(1)-N(10)$ | $26 \cdot 1(2)$ |
|  |  | $C(9)-C(1)-N(10)-C(5)$ | $68.9(2)$ |



Fig. 2. Projection of the structure along $\mathbf{b}$.

Verdier, Florencio \& Garcia-Blanco, 1977) and in $N$-methylgranatanine-3-spiro-5'-hydantoin (Florencio, Smith-Verdier \& Garcia-Blanco, 1978) it can be seen that the deformation in the present compound is smaller than in those compounds; it could be due to the boat conformation adopted by the seven-membered ring that produces a decrease in the interactions between $C\left(3,5^{\prime}\right)$ and the $C(7)$ and $C(8)$ atoms.

The position of the hydantoin ring on the $\mathrm{C}\left(3,5^{\prime}\right)$ atom is similar to that found in $N$-methyltropane3 -spiro- 5 '-hydantoin and in $N$-methylgranatanine-3-spiro-5'-hydantoin.

The molecular packing as viewed along $\mathbf{b}$ is shown in Fig. 2. Hydrogen bonds of types $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}$ link the molecules together. Two $\mathrm{N}\left(1^{\prime}\right) \cdots \mathrm{O}(1)$ bonds of 2.886 (1) $\AA$ and $\mathrm{N}\left(1^{\prime}\right)-\mathrm{H} \cdots \mathrm{O}(1)$ angles of $168(2)^{\circ}$ are formed between pairs of molecules related by a centre of symmetry, while the $\mathrm{N}\left(3^{\prime}\right) \cdots \mathrm{N}(10)$ bond of 2.896 (2) $\AA$ [and $N\left(3^{\prime}\right)-H \cdots N(10)$ angle of $170(2)^{\circ}$ ] links the molecules forming chains along [101]. The other intermolecular contacts correspond to normal van der Waals interactions.

The authors are grateful to the Centro de Proceso de Datos de la Junta de Energia Nuclear, Madrid, for providing facilities for the computations.

## References

Duax, W. L. \& Norton, D. A. (1975). Atlas of Steroid Structure. New York: Plenum.
Florencio, F., Smith-Verdier, P. \& Garcia-Blanco, S. (1978). Acta Cryst. B34, 1317-1321.

International Tables for X-ray Crystallography (1974). Vol. IV, pp. 22-98. Birmingham: Kynoch Press.
Main, P., Fiske, S. J., Hull, S. E., Lessinger, L., Germain, G., Declercq, J. P. \& Woolfson, M. M. (1980). MULTAN 80. A System of Computer Programs for the Automatic Solution of Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
Smith-Verdier, P., Florencio, F. \& García-Blanco, S. (1977). Acta Cryst. B33, 3381-3385.

Stewart, J. M., Kundell, F. A. \& Baldwin, J. C. (1970). The XRAY 70 system. Computer Science Center, Univ. of Maryland, College Park, Maryland.
Vilches, J., Florencio, F., Smith-Verdier, P. \& GarcíaBlanco, S. (1981). Acta Cryst. B37, 2076-2079.


[^0]:    *The Conformation of Heterocyclic Spiro Compounds. XI. Part X: Vilches, Florencio, Smith-Verdier \& Garcia-Blanco (1981).
    $\dagger$ Hydantoin is 2,4-imidazolidenedione.

[^1]:    * Lists of structure factors, anisotropic thermal parameters and deviations of atoms from least-squares planes have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 36766 ( 27 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH 1 2HU, England.

